Scattering factors were taken from International Tables for X-ray Crystallography (1962).

Discussion. The molecule lies in a special position [C(6) on the mirror plane]. The benzene ring is planar. Atoms H(11), atoms O(1) and atoms C(4) are coplanar with the benzene ring, and the methoxy carbon atoms C(7) lie below the benzene plane (0.113 Å). The cycloheptene ring is in the boat conformation in the crystal. Atoms O(2) lie below the benzene plane (0.977 Å) and the angle formed between the C(4)–O(2) bond and the benzene ring is $54 \cdot 16^{\circ}$ (Fig. 1).

There are no intermolecular interactions other than van der Waals.

The torsion angles are given in Table 2.

A full discussion of the conformation of the cycloheptene ring system and a comparison with other such systems will be presented when a study on a group of disubstituted benzocycloheptenediones has been completed.

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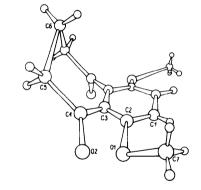


Fig. 1. The observed conformation of the title compound.

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1,4-Dihydroxybenzocycloheptene-5,9(6H,8H)-dione

BY TOMASZ A. OLSZAK, A. STĘPIEŃ, E. WAJSMAN AND M. J. GRABOWSKI

Department of Crystallography, Institute of Chemistry, University of Łódź, 91–416 Łódź, Nowotki 18, Poland

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Abstract. $C_{11}H_{10}O_4$, monoclinic, C2/c, a = 10.637 (2), b = 13.200 (4), c = 6.731 (2) Å, $\beta = 102.55$ (2)°, Z = 4, V = 922.49 Å³, $\mu r = 0.1$. Diffractometer data collected with monochromatic Cu Ka radiation consisted of 867 independent reflections with $I > 2\sigma(I)$. The structure was solved by direct methods and refined by a full-matrix least-squares procedure to a final R value of 0.049. The molecule lies in a special position, the benzene ring is planar and the cycloheptene ring is twisted.

Introduction. Crystals of the title compound were obtained from methanol solution. The average size of the crystals was 0.2-0.3 mm. The absorption coefficient has been neglected because of the small value of $\mu r(\operatorname{Cu} K\alpha) = 0.1$. 931 independent reflections were measured by the $\omega - 2\theta$ step-scan technique with a

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CAD-4 diffractometer and a maximum value of $\sin \theta / \lambda$ of 0.6095 Å⁻¹.

The cell constants were determined with a CAD-4 diffractometer by the least-squares method. 867 reflections were treated as observed $[I > 2\sigma(I)]$ and used to obtain the *E* map (*MULTAN*, Main, Woolfson, Lessinger, Germain & Declercq, 1976). The parameters of 1,4-dimethoxybenzocycloheptene-5,9(6*H*,8*H*)-dione (Olszak, Stępień, Wajsman & Grabowski, 1980) for the non-hydrogen atoms were used to calculate the Debye curve (program *NORMAL*). All non-hydrogen atoms were found on the *E* map and the *R* factor calculated for this model was 0-43. After four isotropic cycles *R* reduced to 0.14. The value of *R* after the next three anisotropic cycles was 0.099. The ΔF map calculated at this stage revealed all the H atoms. These were given isotropic thermal parameters taken from the

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atoms to which they were bonded. Anisotropic thermal parameters were introduced for non-hydrogen atoms and six cycles of refinement were performed. The *ORFLS* program (Busing, Martin & Levy, 1962) was applied. The final *R* factor was 0.049.* The largest peak on the final difference electron density map was $0.4 \text{ e } \text{Å}^{-3}$. The final positional parameters are listed in Table 1 along with the standard deviations. Interatomic distances and angles uncorrected for thermal motion are given in Table 2. Scattering factors were taken from *International Tables for X-ray Crystallography* (1962).

* Lists of structure factors and thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35400 (12 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Final positional parameters $(\times 10^4)$ and B_{iso} $\{B_{iso} = \frac{1}{3}[(4\beta_{11}/a^{*2}) + (4\beta_{22}/b^{*2}) + (4\beta_{33}/c^{*2})]\}$ with standard deviations

	x	,v	z	$B_{\rm iso}/B~({\rm \AA}^2)$
C(1)	5653 (2)	10373 (1)	7614 (3)	3.73 (10)
C(2)	6335 (2)	9450(1)	7711(3)	3.23 (8)
C(3)	5683 (2)	8529 (1)	7555 (3)	2.83 (8)
C(4)	6386 (2)	7614(1)	7169 (3)	3.10 (8)
C(5)	5638 (2)	6732 (1)	6095 (3)	3.59 (9)
C(6)	5000	6090 (2)	7500	4.06 (13)
O(1)	7618 (1)	9517(1)	7861 (3)	4.41 (8)
O(2)	7576(1)	7595 (1)	7567 (3)	4.18(7)
H[C(1)]	6122 (25)	11043 (18)	7711 (44)	2.56 (53)
H O(1)	7935 (32)	8880 (27)	8037 (58)	4.55 (76)
H[1C(5)]	4919 (23)	6989 (15)	4998 (38)	1.06 (36)
H[2C(5)]	6282 (24)	6309 (17)	5535 (41)	2.26 (46)
H[C(6)]	5693 (22)	5624 (17)	8314 (42)	2.19 (45)

 Table 2. Distances (Å) and angles (°) with estimated standard deviations

C(1) C(1*)	1-365 (4)	C(1)C(2)	1-411 (3)
C(1) = H[C(1)]	1.01(2)	C(2)-C(3)	1-393 (3)
C(2) O(1)	1-349 (3)	C(3)-C(3*)	1-439 (4)
C(3) C(4)	1-474 (3)	C(4) - C(5)	1.505 (3)
C(4) O(2)	1.236 (2)	C(5) C(6)	1.534 (3)
C(5)- H[1C(5)]	1.00(2)	C(5) H 2C(5)	1.02(3)
C(6) H C(6)	1.02 (2)	O(1) - H[O(1)]	0.90(4)
O(2) H[O(1)]	1.75 (4)		
C(1*) C(1) C(2)	120.4 (1)	C(4) - C(5) - H[1C(5)]	109 (1)
$C(1^{*}) = C(1) - H[C(1)]$	119 (1)	C(4) $C(5)$ $H[2C(5)]$	106 (1)
C(1) = C(1) = H[C(1)] C(2) = C(1) = H[C(1)]	121 (1)	H[1C(5)] = C(5) = H[2C(5)]	112 (2)
C(1) C(2) C(3)	120.5 (2)	C(6)-C(5)-H 1C(5)	106 (2)
C(1) = C(2) - O(1)	116.7 (2)	C(6) C(5) - H[2C(5)]	110(1)
C(3) = C(2) - O(1)	122.8 (2)	$C(5)-C(6)-C(5^{\bullet})$	112.9 (2)
$C(3^{\bullet}) = C(3) = C(2)$	(19.0(1))	C(5) C(6) H C(6)	107 (2)
$C(3^{*}) C(3) C(2)$	122.5(1)	$C(5)$ - $C(6)$ $H[C(6)^*]$	111 (2)
C(2) = C(3) + C(4)	117.7(2)	H C(6) -C(6)-H C(6)*	106 (3)
C(3) C(4) C(5)	118-9 (1)	C(2) = O(1) = H[O(1)]	107(2)
C(3) = C(4) - O(2)	120.9 (2)	$C(4) \cdot O(2) - H[O(1)]$	101 (1)
C(5) = C(4) = O(2) C(5) = C(4) = O(2)	119.9 (2)	O(1) - H[O(1)] - O(2)	145 (3)
C(4) C(5) C(6)	113.0 (2)		
		C(3) C(4) C(5) C(6)	79-0(2)
$C(1^{\bullet}) C(1) C(2) C(3)$	0.6 (4)	C(3) C(4) C(5) C(6)	43.3 (1)
$C(1) C(2) C(3) C(3^*)$	4.5 (3)	$C(4) C(5) C(6) C(5) - C(4^{\circ})$	43.3(1)
$C(2) = C(3) = C(3^{\bullet}) = C(2^{\bullet})$	6-4 (4)		79.0(2)
$C(3) C(3^*) C(2^*) C(1^*)$	4.5 (3)	$C(6) \cdot C(5^*) \cdot C(4^*) \cdot C(3^*)$	16.7 (3)
$C(3^*) = C(2^*) = C(1^*) - C(1)$	0.6 (4)	$C(5^*)$ $C(4^*)$ - $C(3^*)$ $C(3)$	27.4 (4)
$C(2^{*}) = C(1^{*}) = C(1) = C(2)$	1-4 (5)	$C(4^{\bullet}) \cdot C(3^{\bullet}) \cdot C(3) = C(4)$	16.7 (3)
$C(1^{\bullet})$ $C(1)$ $C(2)$ $O(1)$	177-6 (3)	$C(3^{\bullet}) - C(3) C(4) - C(5)$	10.7(3) 11.3(3)
C(3*) C(3) C(4) O(2)		$O(1) C(2) - C(3) \cdot C(4)$	$21 \cdot 2(3)$
C(6) C(5)-C(4) O(2)	106.7 (2)	C(2)-C(3) C(4)-O(2)	21.2 (3)

* Atoms related by the twofold axis: $x, y, \frac{1}{2} = z$.

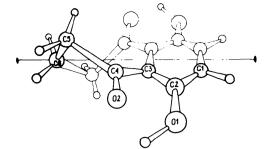


Fig. 1. Observed conformation of the title compound.

Discussion. The molecule of the title compound lies in a special position -C(6) is on the twofold axis.

The benzene ring is almost planar. The best plane taken through the benzene C atoms is given by $\pi \equiv$ -0.982x + 6.678z = 4.517 (x and z are the positional parameters given in Table 1). The average distance of the benzene C atoms from the best plane is ~ 0.03 Å. The torsion angles in the molecule are given in Table 2. There are strong $O(1)-H|O(1)|\cdots O(2)$ hydrogen bonds in the molecule. C(2), C(3), C(4), O(2), O(1) and H[O(1)] form rings which are almost planar. The equation of the best plane taken through C(2), C(3), C(4), O(1), O(2) is $\pi_k \equiv -0.786x - 1.470y + 6.619z$ = 3.229 (x, y and z are the positional parameters given in Table 1). The hydrogen atom H[O(1)] has not been included in the calculation because of the lower precision of its positional parameters. The average distance of the atoms from the best plane π_k is ~ 0.1 Å. H|O(1)| lies 1.6 Å out of the π_k plane. The value of the dihedral angle formed by the planes π and π_k is 152.8°. The cycloheptene ring is in the twisted conformation (Fig. 1).

A full discussion of the conformation of the cycloheptene ring system and a comparison with other such systems will be presented later.

The authors are grateful to Dr Z. Madeja-Kotkowska (Hahn & Madeja-Kotkowska, 1963) for supplying the crystals, and the Polish Academy of Sciences for financial support.

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